

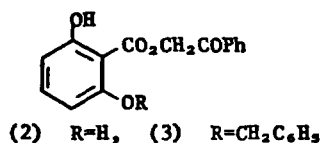
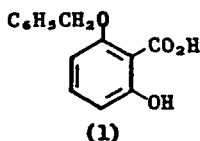
SELECTIVE MONO O-ALKYLATION OF γ -RESORCYLIC ESTERS

by

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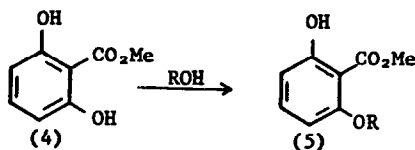
Abstract: The use of diethylazodicarboxylate/triphenylphosphine for the selective mono-O-alkylation of γ -resorcylic esters is described.

No general method has been published for the synthesis of 6-alkoxysalicylic acids. In particular, existing syntheses of 3- and 5-benzyloxysalicylic acids^{1,2} were found to be unsuitable for the preparation of the corresponding 6-substituted isomer (1), a compound we required for the synthesis of zearalanone analogues.³ We now report a method for the synthesis of (1) applicable to a wide range of 6-alkoxy derivatives.



Diethylazodicarboxylate has been used in the synthesis of aryl alkyl ethers.⁴ The mild, neutral conditions associated with this reagent lead to selective alkylation of resorcylic esters. Phenacylresorcyate⁵ (2) in dry tetrahydrofuran containing triphenylphosphine (1.5eq) and benzyl alcohol (1.5eq) at 25°C under an atmosphere of nitrogen was treated with a solution of diethylazodicarboxylate (1.5eq) in dry tetrahydrofuran added dropwise over a period of 2 hours. The solvent was evaporated leaving a gum which was triturated with cold ether to remove insoluble by-products. The ether was then evaporated to give phenacyl 6-benzyloxysalicylate (3) which was recrystallised from isopropyl alcohol (70%, m.p. 119°C). Treatment of (3) with zinc in acetic acid gave the desired acid (1) in 50% overall yield (m.p. 121-2°C, from isopropyl alcohol).

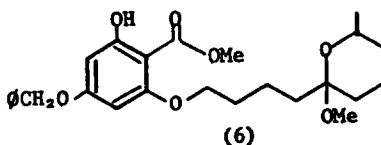
To test the generality of our procedure, a range of alcohols have been employed to alkylate the simpler methyl ester (4). The results are summarised below.



R	Me	Et	iPr	Allyl	benzyl
% yield ^{a, b}	76	61	55	62	73
m.p. (°C)	40-42	46	-	-	-
b.p. (°C, 0.3mm Hg)	-	-	80	65	-

- a. correct microanalytical data were obtained for all novel compounds.
- b. products obtained as gums were purified by chromatography on Whatman S13 TLC silica followed by distillation using a ball tube oven.

The value of this new procedure was further demonstrated by the preparation in 60% yield of the key intermediate (6) used in the synthesis of a macrocyclic lactone.³



References

1. S.C. Datta, V.V.S. Murti, and T.R. Seshadri, Indian J. Chem., 1969, **7**, 110.
2. H. Baganz and R. Krattner, Arch. Pharm., 1960, **293**, 393.
3. R.J. Bass, B.J. Banks, M.R.G. Leeming and M. Snarey, submitted for publication in J. Chem. Soc., Perkin I.
4. M.S. Manhas, W.H. Hoffman, Bansi Lai, and A.K. Bose, J. Chem. Soc., Perkin I, 1975, 461; S. Bittner and Y. Assaf, Chem. and Ind., 1975, 281.
5. Prepared from phenacylbromide and γ -resorcylic acid using potassium hydrogen carbonate in dry dimethylformamide (80%, m.p. 67°C from methanol).

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